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Proteasome Inhibition by a Totally Synthetic β -Lactam Related to Salinosporamide A and Omuralide

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Salinosporamide A (1) and omuralide (2) are potent naturally derived substances which inhibit proteasome function with very high selectivity.^{1–3} Proteasome inhibition offers considerable promise in the therapy of a number of types of cancer and is already used for multiple myeloma.⁴ Several routes have been developed for the syntheses of 1 and 2.^{1.5} One potential problem with the use of 1 or 2 as therapeutic agents is their short half-life in solution at pH 7 or in serum (estimated as 5-10 min). Because of this potential shortcoming, we have developed a synthesis of the β -lactam 3, which is expected to be much more stable than the corresponding β -lactone (or 1 and 2).

The pathway of the synthesis is outlined in Scheme 1. The known alcohol 4^{6a} was converted to oxazolidinone 5^{6b} (93% yield) by successive acylation with phenyl chloroformate and pyridine in dichloromethane and reduction of the azide function with concomitant cyclization. N-Protection of 5 with *tert*-butylpyrocarbonate– 4-(dimethylamino)pyridine (DMAP)–Et₃N afforded 6. Reductive cleavage of the benzyl ester subunit in 6 provided the carboxylic acid 7.

Addition of a solution of **7** to a solution of 1-chloro-N,N-2-trimethylpropenylamine⁷ led to smooth formation of the corresponding acid chloride. Rapid addition of the acid chloride to a mixture of the imine **8**⁸ and triethylamine gave stereoselectively

the β -lactam 9 in 43% yield. Noteworthy in this transformation is the use of an imine derived from (S)-(-)-1-(4-methoxyphenyl)ethylamine. The corresponding imine derived from 4-methoxybenzylamine led to β -lactam but in lower yields (~12%). Cleavage of the oxazolidinone ring in 9 was accomplished by exposure to Cs₂-CO₃-MeOH to afford the alcohol 10 in high yield (94%).⁹ Oxidation of the 2-trimethylsilylfuran subunit in 10 by peroxyacetic acid gave the butenolide 11.10 Catalytic reduction of 11 provided the butyrolactone 12 in 91% yield, the remainder of the material being the corresponding diastereomer. Cleavage of the t-butoxycarboxyl group of 12 was accomplished with trimethylsilyl trifluoromethanesulfonate and 2,6-lutidine in dichloromethane at 23 °C. Any remaining trimethylsilyl trifluoromethanesulfonate was quenched by addition of methanol and triethylamine to the reaction mixture. Fluoride treatment during workup provided the desired butyrolactam 13 in 86% yield. The chlorine atom was introduced by selective reaction of the primary hydroxyl of 13 with dichlorotriphenylphosphorane and pyridine in acetonitrile at 23 °C to give 14 in 80% yield. Oxidative cleavage of the (S)-1-(-)-(4-methoxyphenyl)ethylamide protecting group in 14 with ceric ammonium nitrate (CAN) cleanly afforded the β -lactam 3 in 89% yield. We were pleased that **3** is completely stable at pH 7 and 23 °C for 24 h.

Scheme 1



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Figure 2. Plot of ln(fluorescence/time) versus time during proteasome inactivation experiments. The slope of these lines are equivalent to k_{obs} for the inactivation of 20S proteasome by the β -lactam 3. The half-lives of inactivation of 20S proteasome by 3 are approximately 58 and 165 min at concentrations of 3 at 30 and 10 μ M, respectively.

Determination of the inhibitory activity of 3 against the 20S proteasome12 was carried out by a modification of the method reported by McCormack and Dick et al.¹¹ Incubation of 20S proteasome (0.5 μ g/mL) with 3 at 10 and 30 μ M in pH 7.9 buffer (20 mM Hepes, 0.5 mM EDTA, 0.025% w/v SDS) at 37 °C resulted in a time-dependent loss in the proteasome's ability to hydrolyze Suc-Leu-Leu-Val-Tyr-AMC (50 μ M).¹² The progress of proteasome inactivation was determined by addition of Suc-LLVY-AMC to a 2 mL aliquot of the incubation experiment and monitoring the rate of production of free AMC by fluorescence spectrophotometry. A plot of ln(fluorescence/time) versus time provided a linear decay the slope of which gave k_{obs} for proteasome inactivation (Figure 2). The rate of proteasome inactivation was proportional to the concentration of 3 present in the incubation experiments. The 10 and 30 μ M time course experiments gave values for $k_{obs}/[3]$ as 7 and 6.7 M⁻¹ s⁻¹, respectively. To monitor background inactivation of 20S proteasome,¹³ each of these of experiments was run side by side with a control experiment identical in all respects except for the exclusion of 3. In the control experiment correlating to $30 \,\mu\text{M}$ of 3, after 3 h, \sim 30% of initial proteasome activity was lost. In the control experiment correlating to 10 μ M of 3, after 9.5 h, ~50% of the initial proteasome activity remained. In each the corresponding experiments containing 3, less than 10% of the initial proteasome activity remained at these respective time points. Although the rate of proteasome inhibition by 3 is considerably slower than

that for 1 or 2, the slow rate is more than compensated for by the greatly increased aqueous stability of 3 under physiological conditions.

It seems reasonable that the pathway of proteasome inhibition by the β -lactam **3** follows that of omuralide and salinosporamide A, that is, acylation of a catalytically active threonine of a proteolytic β -subunit. It is likely also that this acylation is rendered irreversible by ring closure involving the chloroethyl group as an electrophile, as appears to be the case for salinosporamide A,^{5b} since treatment of 3 with methanolic base afforded the bicyclic pyrrolidine 15. This fact, the observation of proteasome inhibition in vitro, and the indefinite stability in neutral aqueous solution suggest that 3 is a worthy candidate for further biological evaluation.

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Supporting Information Available: Experimental synthetic procedures as well as physical and spectral data for compounds 3 and 5-15, experimental procedures for incubation of 20S proteasome with 3, and determination of inactivation rates. This material is available free of charge via the Internet at http://pubs.acs.org.

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